

II. CHEMICA

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NUTRITIONAL REQUIREMENTS OF  
LACTIC ACID BACTERIA

I. THE CALCIUM REQUIREMENTS OF  
STREPTOCOCCUS THERMOPHILUS  
STRAINS

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STRAINS

## Nutritional requirements of lactic acid bacteria

From the standpoint of growth factor requirements lactic acid bacteria are among the most complex of the micro-organisms that have been investigated. The growth factors which either stimulate or are necessary for growth of lactic acid bacteria include various amino acids, peptides, vitamins of the B complex (and related compounds), purine and pyrimidine bases and their derivatives, and fatty acids. The known growth factor requirements of these bacteria have been discussed in numerous reviews <sup>1-7</sup>. Certain strains of lactic acid bacteria, including those which are commonly used in the dairy industry (e.g. the strains *Streptococcus thermophilus*, *Lactobacillus helveticus*, *Lactobacillus lactis*, and *Lactobacillus bulgaricus*) require unknown factors for growth in synthetic media. A more complete knowledge of the nutritional requirements of the lactic acid bacteria used in cheese manufacture is therefore a necessary foundation for any improvements in the quality of cheese.

Advantage can be taken of the fact that lactic acid bacteria require unknown organic growth factors to detect, isolate and identify such factors in biological materials. Until now, however, this possibility of discovering new growth factors by means of lactic acid bacteria used in cheesemaking has been very little utilized. The first aim of this work was to find basal media in which the growth of the lactic acid bacteria would be optimal and which could be used in attempts to isolate new growth factors. It was necessary, first of all, to find out in what way the different chemical compositions of the growth media affect the growth of lactic acid bacteria.

## Experimental

### *Cultures and their maintenance*

The experiments were carried out on 124 strains of lactic acid bacteria which had been isolated as pure cultures from various Finnish dairy products (milk, butter, buttermilk, cheese). The strains were kindly supplied by

Professor A. I. Virtanen, Director of the Biochemical Institute, Helsinki. These lactic acid bacteria represent a large group of bacteria whose nutritional requirements have not yet been investigated. Some of the strains are commonly used in the manufacture of cheese in Finland.

Cultures of the organisms were maintained as stab cultures in test tubes containing agar medium (TSHGA- agar medium) or sterile skim milk. The TSHGA- agar medium contained 1 per cent glucose, 0.5 per cent lactose, 0.5 per cent yeast extract, 0.5 per cent sucrose, 0.5 per cent Bacto-tryptone, gelatine 0.25 per cent, ascorbic acid 0.07 per cent, and 1.5 per cent agar. The pH was adjusted to 6.7–6.8. The frequency of transfer varied from once a fortnight to once a month. The cultures were maintained at 4° between transfers.

#### *Preparation of inoculum cultures*

The composition of the inoculum medium was that described above but without the agar. This medium was divided into 7-ml aliquots which were sterilized in an autoclave 10–15 min. at 112° and then kept in a refrigerator. The inocula were prepared by transferring the organism from the stab culture to this medium. After being incubated at 37° for 12–48 hrs, depending on the strain in question, the cells were centrifuged and washed with 5–10 ml of 0.9 per cent sodium chloride solution. The centrifugation and washing were repeated three times. Finally the cells were suspended in 50 ml of saline. One drop of the slightly opaque suspension was used to inoculate the medium in each tube. Growth experiments were carried out at 37 or 42°. In general, the incubation was stopped when the logarithmic growth phase was reached in order to make possible a comparison of the rates of growth of different strains.

#### *Compositions of basal media*

The compositions of the basal media used in the investigation are shown in Table 1. All the media are modifications of the growth medium for lactic acid bacteria described by Anderson and Elliker<sup>8</sup>. They were prepared by mixing stock solutions of the various compounds. Separate solutions of vitamins, purine and pyrimidine bases were prepared, but the inorganic salts were combined in a single solution unless stated otherwise. The basal media were adjusted to pH 6.7 before being used. All stock solutions were stored in a refrigerator and replaced monthly.

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Table 1.

The compositions of basal media (quantities per 100 ml).

Component	Quantity	S	SII	SIII	SIV	BI	BII	CI	CII	CIII	CIV
Glucose	mg	1000	1000	1000	1000	1000	1000	1000	1000	1000	1000
Lactose	»	500	500	500	500	500	500	500	500	500	500
Sucrose	»	500	500	500	500	500	500	500	500	500	500
DL-Aspartic acid	»	20	20	20	20	20	20	20	20	20	20
L-Glutamic acid	»	10	10	10	10	10	10	10	10	10	10
Sodium acetate	»	2000	2000	2000	2000	—	1000	—	1000	34	340
Xanthine	»	2	2	2	2	2	2	2	2	2	2
Casamino acids	»	1000	1000	500	500	500	500	500	500	500	500
Tween 80	»	100	100	100	100	100	100	100	100	100	100
Oleic acid	»	1	1	1	1	1	1	1	1	1	1
DL-Tryptophan	»	20	20	20	20	20	20	20	20	20	20
L-Cystine	»	10	10	10	10	10	10	10	10	10	10
Thymine	µg	100	100	100	100	100	100	100	100	100	100
Cytosine	»	100	100	100	100	100	100	100	100	100	100
Hypoxanthine	»	100	100	100	100	100	100	100	100	100	100
Adenine, Guanine, and Uracil, each	mg	2	2	2	2	2	2	2	2	2	2
Ascorbic acid	»	40	40	20	20	20	20	20	20	20	20
Thiamine HCl	µg	100	100	50	50	50	50	50	50	50	50
Riboflavin	»	100	100	50	50	50	50	50	50	50	50
Nicotinic acid	»	100	100	50	50	50	50	50	50	50	50
P-Aminobenzoic acid	µg	100	100	50	50	50	50	50	50	50	50
Ca-pantothenate	»	100	100	50	50	50	50	50	50	50	50
Biotin	mµg	100	100	50	50	50	50	50	50	50	50
Vitamin B <sub>12</sub>	»	22	22	11	11	11	11	11	11	11	11
Folic acid	»	1000	1000	500	500	500	500	500	500	500	500
Pyridoxal phosphate	»	400	400	200	200	200	200	200	200	200	200
Pyridoxaminephosph.	µg	10	10	5	5	5	5	5	5	5	5
Pyridoxamine HCl	»	10	10	5	5	5	5	5	5	5	5
Pyridoxal	»	20	20	10	10	10	10	10	10	10	10
Pyridoxine HCl	mg	4	4	2	2	2	2	2	2	2	2
KH <sub>2</sub> PO <sub>4</sub>	»	1000	1000	—	—	—	—	—	—	12	—
FeSO <sub>4</sub> · 7 H <sub>2</sub> O	µg	4000	4000	1330	400	400	400	400	400	400	400
MnSO <sub>4</sub> · 4 H <sub>2</sub> O	mg	13	13	3	0.4	0.4	0.4	0.4	0.4	0.4	0.4
MgSO <sub>4</sub> · 7 H <sub>2</sub> O	»	80	80	40	30	30	30	30	30	30	30
KCl	»	—	—	550	140	140	140	140	140	140	140
NaCl	»	4	4	—	—	—	—	4	4	4	4
Glycero-2-phosphate	»	—	—	78	—	156	156	—	—	468	468
Sodium citrate · 2 H <sub>2</sub> O	»	500	—	—	—	—	—	—	—	—	—
Succinic acid	»	—	—	—	—	—	—	—	—	295	295

Medium SIIIb is the same as SIV but contains also 156 mg of glycero-2-phosphate per 100 ml.

All media contained traces of ammonia salt because xanthine was dissolved in strong ammonia.

Sodium hydroxide or hydrochloric acid was used to neutralize the media to pH 6.8 and therefore the concentrations of Na<sup>+</sup> or Cl<sup>-</sup> ions in the media are somewhat higher than the table indicates.

### General procedure

In principle the experimental procedure used in the study was the same as that used in this laboratory for the determination of vitamins and amino acids. Two and a half millilitres of the basal medium was added to a test tube and diluted with water or supplements to 5 ml. The test tubes were then plugged with cotton wool and autoclaved 5–10 min. at 112–115°. After being cooled, the tubes were inoculated and then incubated at 37 or 42°. The growth of the lactic acid bacteria was followed turbidimetrically, the extent of growth being recorded as galvanometer readings of a Klett-Summerson colorimeter with a 590–660  $m\mu$  filter. In the tables showing the results, the rates of growth are expressed directly as galvanometer readings.

### Results

In the initial stage of the work, several of the 124 strains of lactic acid bacteria tested appeared to possess very complex growth factor requirements. It also became clear during the first experiments that some strains of *Streptococcus thermophilus* require calcium for growth when synthetic media are used. Because it is very important to know the exact nutritional requirements of these calcium-requiring strains, the next step was a detailed investigation of the effect of different inorganic and organic components of the basal media on the growth of these bacteria. The results of these experiments are shown in Tables 2–25. It should be noted that these results relate to strains which are sensitive to calcium. About 30 strains (mainly *Streptococcus thermophilus*) of the 124 strains preliminarily tested were sensitive in this respect.

*Potassium.* The effect of potassium on certain strains is shown in Table 2.

**Table 2.**  
Effect of potassium concentration.

Basal medium	Strain	mg. of K <sup>+</sup> /ml						Incubation at 42°C, hours
		0	0.7	1.4	2.1	2.9	4.3	
SIII	Str 10	—	150	110	40	—	—	24
»	Str 11	—	110	100	90	60	—	24
»	Str 110	—	150	110	90	—	—	24
»	Str 111	—	170	160	160	140	20	24
»	Cas I	—	150	170	130	60	—	70
»	Cas J (7)	—	200	230	325	—	—	70
»	Kungsbölen k.	—	190	190	190	180	150	24
»	Kauhajoen k.	—	190	170	120	—	—	24

As can be seen, potassium ions are required by all the strains. Already the lowest concentration of potassium (37  $\mu$ moles per ml) effected maximal growth under the experimental conditions.

*Manganese.* The effect of manganese on the growth of 10 strains is seen in Table 3. The basal media used in these experiments were SII and SIII, but without the manganese sulphate. The results show that manganese ions are not generally required by these strains.

Table 3.  
Effect of manganese concentration.

Basal medium	Strain	$\mu$ g Mn <sup>++</sup> /ml							Incubation at 42°C, hours
		0.0	1.0	3.0	5.0	7.5	12.5	25.0	
SII	Str 10	125			130		100	100	18
»	Str 10	180			180		170	170	40
SIII	Str 10	75	77	82	88	91	90		24
»	Str 11	55	77	76	79	79	79		24
»	Str 110	155	160	148	146	142	140		24
»	Str 111	175	172	168	162	161	168		24
»	Kauhaj. k.	179	174	172	160	178	165		24
»	Kungsby. k.	210	201	198	195	200	191		24
»	Cas J (7)	20	144	115	70				24
»	Cas I	146	170	146	165	57	29		24
SII	» »	145			125	65	54	50	18
»	» »	175			215	75	25	33	24
»	» »	235			290	280	260	105	40
»	Ylilied. k.	—			115	125	98	108	18
»	» »	—			115	126	120	110	24
»	» »	166			183	185	166	179	40
»	Lb helv. 1182	185			185	183	155	12	60

*Iron.* Table 4 shows the effect of iron on the growth of 9 strains. Iron (II) ions do not seem necessary for rapid growth under the experimental conditions used. With some strains, iron was observed to promote growth.

*Magnesium.* The magnesium requirements of 10 strains were tested. As can be seen from Table 5, magnesium ions are essential for the growth of most strains and stimulate the growth of the other strains.

*Citrate.* The preliminary experiments showed that sodium citrate inhibits the growth of certain strains of lactic acid bacteria (named Group II). These findings were confirmed and the experiments were extended to an

Table 4.

Effect of iron concentration.

Basal medium	Strain	$\mu\text{g Fe}^{++}/\text{ml}$							Incubation at 42°C, hours
		0.0	0.8	1.6	3.2	4.0	4.8	8.0	
SIII	Str 10	57	67	54	40		35	26	24
SII	Str 10	110	158	188	155	142		90	24
»	Str 10	140	165	198	150	160		180	45
SIII	Str 11	38	69	63	58		50	74	24
SII	Str 111	210	165	250	190	186		180	45
SIII	Str 111	155	148	157	145		136	136	24
»	Kauh. k.	165	169	161	161		144	122	24
»	Kungsb. k.	178	181	172	172		150	150	24
»	Str 110	138	132	119	113		—	—	24
»	Cas J (7)	10	96	115	100		117	95	24
»	Cas I	145	187	187	199		170	132	24
SII	» I	—	97	135	105		110	114	27
»	» »	—	115	115	97		78	50	45
»	Lb helv.								
	1182	20	78	129	206		200	190	45

Table 5.

Effect of magnesium concentration.

Basal medium	Strain	$\mu\text{g Mg}^{++}/\text{ml}$						Incubation at 42°C, hours
		0.0	20.0	40.0	60.0	80.0	120.0	
SII	Str 10	—	125	125	120	107		27
SIII	Str 10	—	105	106	103	90	95	24
SII	Str 10	—	153	174	178	178		40
SIII	Str 11	47	106	112	100	100	92	24
»	Str 111	—	170	167	167	165	160	24
»	Kauh. k.	—	170	166	155	152	145	24
»	Kungsb. k.	—	194	195	198	206	209	24
»	Cas J (7)	—	115	230	226	210	140	72
»	Str 110	—	128	111	138	46	11	24
SII	Cas I	—	15	—	10	55		24
»	» »	30	230	220	270	185		40
SIII	» »	12	185	192	189	174	173	72
SII	Ylilied. k.	—	130	135	120	110		24
»	» »	167	184	185	166	179		40
»	Lb helv. 1182	5	24	16	14	47		64



examination of the tolerance of citrate by 25 strains of Group II using basal medium SIII (or SII). The citrate tolerance varied considerably with different strains. The strains were divided into four groups according to the degree of tolerance (Table 6).

Table 6.  
Tolerance of citrate by different strains of lactic acid bacteria.

Group IIA, tolerates less than 1 mg	Group IIB, tolerates 1–2 mg	Group IIC, tolerates 2–3 mg	Group IID, tolerates more than 3 mg
Str 11 Str 40 Str 75 Str 100 Str 102 Str 103 Str 110 Ths <sup>1)</sup> Kauhaj.k. Ylilied.k. <sup>2)</sup>	Str 10 Str 11R Str 72 Str 74 Str 101 Str 111 <sup>1)</sup> Cas I <sup>2)</sup> Cas J (7) H.th.str	Hatt.s. Lb lact 1183 B.c.h.H.	Lb 80 Lb 83 Lb helv 1182 <sup>1)</sup>

<sup>1)</sup> Strain tested in medium SII.

<sup>2)</sup> Strain tested in media SII and SIII.

Table 7.  
Effect of calcium.

Strain	$\mu\text{g Ca}^{++}/\text{ml}$									
	—	1	2	3	4	5	10	15	20	30
Str 10	5	18	6	2	95	128	125	135	130	130
Str 10	3	14	—	6	—	115	135	—	125	140
Str 11	1	10	12	11	6	11	22	52	93	115
Str 11	6	19	5	3	6	7	11	41	26	85
Str 11R	9	18	45	82	105	108	142	140	141	139
Str 11R	7	19	110	100	135	140	139	137	139	140
Str 110	—	15	18	178	142	175	230	200	228	225
Str 110	—	14	92	175	184	182	225	214	200	190
Cas I	39	—	39	63	94	117	—	195	240	244
» »	40	280	65	66	97	127	—	200	240	252

Basal medium S, in which the concentration of sodium citrate was 2.5 mg per ml, was used. Incubation 48 hrs at 42°.

*Calcium.* In a medium containing citrate, calcium is either necessary for or a stimulus to the growth of most strains. In a citrate-free basal medium, calcium has no effect on the growth in most cases (Table 8). However, as

Table 8.

Effect of calcium on growth in media containing citrate (SIVSS) and in citrate-free media (SIV). Incubation at 42°.

Strain	SIVSS $\mu\text{g Ca}^{++}/\text{ml}$				Hours	SIV $\mu\text{g Ca}^{++}/\text{ml}$				Hours
	0	5	10	15		0	5	10	15	
Str 10	—	—	30	30	e 24	50	50	45	45	o 24
Str 10R	—	—	—	—	72	—	15	40	50	e 24
Str 11	—	35	45	45	e 24	50	50	55	55	o 24
Str 11R	—	—	—	—	72	—	50	40	40	e 48
Str 12	—	—	40	80	e 72	—	15	45	60	e 24
Str 40	—	40	50	60	e 48	60	60	60	65	o 48
Str 72	—	—	90	100	e 24	—	80	100	105	e 24
Str 74	—	50	80	75	e 48	70	70	70	75	o 24
Str 75	—	40	45	55	e 48	70	60	60	70	o 48
Str 77	—	—	75	130	e 72	—	30	60	80	e 24
Str 100	—	—	45	45	e 48	55	60	65	60	o 24
Str 101	—	—	50	30	e 48	—	5	20	20	e 24
Str 102	—	20	60	60	e 48	35	45	45	50	o 24
Str 103	—	—	60	65	e 48	40	55	60	55	o 24
Str 110	—	80	145	115	e 24	155	120	130	135	o 24
Str 111	—	80	120	105	24	140	125	120	105	o 24
Th	—	—	45	80	e 24	70	105	110	115	s 24
Ths	—	5	30	105	e 24	—	50	110	110	e 24
Cas I	—	40	100	110	e 24	110	125	125	120	o 24
Cas J (7)	5	30	100	140	s 24	130	140	140	140	o 24
H.th.str.	—	—	10	5	e 48	—	10	30	30	e 24
Hatt.s.	20	55	75	75	s 24	140	125	120	105	o 24
Kauh.k.	—	130	150	160	e 48	145	160	155	155	o 48
Kungsb.k.	110	50	100	130	48	120	120	130	120	o 48
Lb 83	135	135	125	100	o 48	150	130	125	130	o 48
Lb lact. 1183	10	130	125	105	s 48	125	130	135	135	s 48

Notes: e = calcium essential for growth.

s = calcium stimulates growth.

o = calcium has no effect on the growth.

can be seen from Tables 9 and 12, the composition of the basal media also otherwise affects the calcium requirements of the strains. On the basis of their calcium requirements in the SIV and SIVSS basal media, the strains were divided into three groups as shown in Table 10. These groups are:

Table 9.

Effect of calcium on the growth in media SII and SIIb lacking citrate.  
Incubation at 42°.

Strain	SIIb $\mu\text{g Ca}^{++}/\text{ml}$				Hours	SII $\mu\text{g Ca}^{++}/\text{ml}$						Hours
	0	5	10	15		0	5	10	15	20	25	
Str 10	50	70	65	65	o 24	125	115	110	105	105	105	o 24
Str 10R	—	5	30	45	e 24							
Str 11	45	60	60	60	o 24	80	80	80	80	90	80	o 24
Str 11R	45	55	55	60	o 24	110	105	100	90	90	90	o 24
Str 12	—	5	20	35	e 24							
Str 40	70	70	70	70	o 24	100	115	110	110	110	110	o 24
Str 72	25	120	130	130	s 24	—	20	30		100	100	e 24
Str 74	85	85	75	75	o 24	—	120	120	120	120	125	e 24
Str 75	60	65	60	55	o 24	—	20	100	80	70	110	e 24
Str 77	—	20	65	85	e 24	—	—	—	—	5	15	e 24
Str 100	70	70	70	70	o 24	—	—	5	20	25	20	e 24
Str 101	40	50	50		o 24	—	20	35	55	50	50	e 24
Str 102	25	50	55	60	s 24	—	—	5	5	5	10	e 24
Str 103	45	70	70	65	o 24	—	5	10	35	50	45	e 24
Str 110	160	120	120	125	o 24	140	155	155	165	160	160	o 24
Str 111	145	130	135	135	o 24	190	205	205	210	210	215	o 24
Th	5	10	115	140	s 24	—	15	100	135	150		e 24
Ths	150	160	175	190	s 24	—	35	125	145	150	155	e 24
Cas I	20	125	125	125	s 24	185	290		320			s 24
Cas J (7)	170	180	200	195	o 24	—	140	180	250	290	270	e 24
H.th.str	—	45	60	60	e 24	—	—	5	5	5	10	e 24
Hatt.s.	180	180	190	180	o 24							
Kauh.k.	125	125	135	130	o 24	185	190	190	190	190	185	o 24
Kungs.b.k.	135	140	140	145	o 24	185	185	185	185	185	175	o 24
Ylilied.k.	45	45	35	35	o 24	—	—	40	50	60	65	e 24
Lb 80	175	175	190	170	o 24	220	220	250	245	245	225	o 24
Lb 83	105	105	100	105	o 24	320	310	330	320		320	o 24
Lb helv. 1182						200	205	150				o 24
Lb lact. 1183	90	165	195	200	s 24	50	80		130	140		s 24

Notes: e = calcium essential for growth.  
s = calcium stimulates growth.  
o = calcium has no effect on the growth.

IIa: strains requiring calcium ions in both media,

IIb: strains requiring calcium ions only in the medium containing citrate, and

IIc: strains not requiring calcium ions in either medium.

In addition, the first two groups can be divided into subgroups.

IIa1 and IIb1: calcium ions are essential for growth, and

IIa2 and IIb2: calcium ions stimulate the growth.

Similar groups can be formed on the basis of the experiments carried out in the basal media SII and SIIIb — these media differ from each other primarily in phosphate content (Table 11). The groups are in this case:

IIx1: the strains require calcium ions in both media,

IIx2: calcium ions stimulate the growth in SIIIb medium,

IIy2: calcium ions are not necessary for growth in medium SIIIb but stimulate the growth in medium SII, and

IIz: the strains do not require calcium ions for growth in either medium.

These results show that the importance of calcium ions is not restricted to a counteraction of the growth inhibition by citrate. It seems probable that phosphate in the concentrations employed also exerts an inhibiting effect on several strains. Calcium ions seem to counteract this inhibition as well.

The experiments proved that calcium counteracts the growth inhibition by citrate. The next step was to study possible counteracting effects of the other components of the basal media.

Table 10.

Grouping of strains according to the tests carried out with calcium in media SIV and SIVSS.

IIa1	IIa2	IIb1	IIb2	IIc
Str 10R Str 11R Str 12 Str 72 Str 77 Str 101 Ths H.th.str.	Th Lb lact 1183	Str 10 Str 11 Str 40 Str 74 Str 75 Str 100 Str 102 Str 103 Str 110 Str 111 Cas I Kauhaj.k.	Cas J (7) Hatt.s.	Kungsbo.k. Lb 83

Table 11.

Grouping of strains according to tests carried out with calcium in the media  
SII and SIIIb.

IIx1	IIx2	IIy1	IIy2	IIz
Str 10R Str 12 Str 77 H.th.str.	Str 72 Str 102 Th Ths Cas I Lb lact 1183	Str 74 Str 75 Str 100 Str 101 Str 103 Cas J (7) Ylilied.k.		Str 10 Str 11 Str 11R Str 40 Str 110 Str 111 Hatt.s. Kauh.k. Kungs.b.k. Lb 83

Table 12.

Effect of calcium in the medium CIV containing 2.5 mg of sodium citrate per ml.  
Incubation 45 hours at 42°.

Strain	Mμmole Ca <sup>++</sup> /ml:	—	25	50	75	100
	μg Ca <sup>++</sup> /ml:	—	6	12	19	25
Ths		—	50	88	98	90
Str 12		—	—	—	—	—
Str 10		—	—	5	62	71
Th		—	—	13	13	87

*Phosphate.* In the experiments concerning calcium requirements it appeared that high phosphate concentrations inhibited the growth of several strains. Phosphate seemed to have no effect on the growth in calcium-free media when used in low concentrations, but inhibition of growth was soon observed when the concentration was increased. Calcium ions counteracted this inhibition (Tables 13, 14 and 15).

*Acetate.* As can be seen from Table 16, a much stronger growth was observed in a medium containing both calcium and acetate ions than in a medium containing acetate ions, but no calcium.

Table 13.

Effect of phosphate.

Basal medium	Strain	$\mu\text{mole/ml}$								Incub. hours	
		0	15	20	30	40	60	75	80		
CICaTa <sup>1)</sup>	Str 10	—	—	—	—	—	—	—	—	24	K <sup>4)</sup>
CIITa <sup>2)</sup>	Str 10	80	—	70	—	50	40	10	—	24	
CII	Str 10	40	—	30	—	10	—	—	—	24	
CIIB <sup>3)</sup>	Str 10	80	—	80	—	70	60	40	—	24	Na <sup>5)</sup>
CI	Str 10	—	—	48	—	128	14	—	19	48	
»	Str 10	5	—	141	—	200	—	—	—	48	
CIV	Str 10	—	—	—	—	6	6	—	6	48	»
»	Str 10	30	—	—	—	—	—	—	—	48	
CI	Str 74	—	—	4	—	—	—	—	—	48	
»	Str 74	—	—	59	—	13	11	—	15	48	»
CIV	Str 74	179	—	—	—	—	—	—	—	48	
»	Str 74	—	—	—	—	—	—	—	—	48	»
CI	Str 75	0	—	3	—	43	13	—	13	48	
»	Str 75	15	—	109	—	—	—	—	—	48	»
CIV	Str 75	—	—	—	—	—	—	—	—	48	
CII	Str 77	70	—	70	—	—	—	—	7	48	K <sup>4)</sup>
CIIB <sup>3)</sup>	Str 77	—	—	—	—	40	10	—	—	24	
CI	Str 101	—	—	37	—	—	—	—	—	24	
»	Str 101	9	—	104	—	—	222	—	22	48	Na <sup>5)</sup>
CIV	Str 101	—	—	5	—	—	—	—	—	48	
CICaTa <sup>1)</sup>	Hatt.s.	10	—	—	—	—	—	—	—	48	»
CIITa <sup>2)</sup>	Hatt.s.	170	—	230	—	200	160	150	—	24	
CII	Hatt.s.	170	—	270	—	280	270	250	—	24	»
CIIB <sup>3)</sup>	Hatt.s.	190	—	290	—	280	280	280	—	24	
CI	Cas J (7)	68	—	134	—	222	278	—	22	48	Na <sup>5)</sup>
»	Cas J (7)	80	—	198	—	—	—	—	—	48	
CIV	Cas J (7)	270	—	300	—	300	7	—	6	48	
»	Cas J (7)	261	—	—	—	—	—	—	—	48	»
CICaTa <sup>1)</sup>	Lb 80	120	210	—	190	—	90	—	—	24	
CIITa <sup>2)</sup>	Lb 80	150	—	300	—	300	300	300	—	24	
CII	Lb 80	150	—	300	—	300	300	300	—	24	»
CIIB <sup>3)</sup>	Lb 80	150	—	300	—	300	300	300	—	24	
										24	

<sup>1)</sup> CI + 5  $\mu\text{g}$  Ca<sup>++</sup>/ml + 100  $\mu\text{mole}$  tartaric acid/ml.

<sup>2)</sup> CII + 50  $\mu\text{mole}$  tartaric acid/ml.

<sup>3)</sup> CII + 50  $\mu\text{mole/ml}$  each of malic acid, malonic acid, succinic acid, and tartaric acid.

<sup>4)</sup> The phosphate was added in the form of primary potassium phosphate.

<sup>5)</sup> The phosphate was added in the form of primary sodium phosphate.

Table 14.

Effect of potassium phosphate in medium CI containing acetate, glycerophosphate, succinate and tartrate (5  $\mu$ mole/ml of each) tested with and without 5  $\mu$ g  $\text{Ca}^{++}$ /ml. Incubation at 42°.

Strain	$\mu$ mole/ml						Incubation hours
	0	5	10	20	50	100	
Without calcium:							
Th	72	78	62	48	12	—	24
Str 77	70	57	61	85	—	—	70
Str 10	50	55	38	35	11	7	24
Hatt.s.	107	94	88	79	79	61	24
Lb. 80	—	184	194	181	194	190	70
With calcium:							
Th	134	129	145	120	100	49	24
Str 77	64	76	82	79	62	22	24
Str 10	66	58	63	49	34	5	24
Hatt.s.	92	103	137	114	140	73	24
Lb 80	175	181	195	199	220	198	70

*Malate, malonate, succinate, tartrate and glycerophosphate.* As very little information is available on the effect of these organic acids on the growth of lactic acid bacteria, a series of experiments were carried out with these acids. The preliminary findings are shown in Tables 17—23. These findings still require checking, but they show clearly that succinate in low concentration stimulates the growth of certain strains in the absence of calcium ions but stimulates the growth even more when calcium ions are present.

Table 24 shows results obtained in experiments where the effect of citrate was studied in a basal medium lacking acetate but containing other organic anions. Also these findings are preliminary, but it is seen that growth inhibition by citrate is stronger when organic anions other than acetate are present in the medium.

*Other organic compounds.* Table 25 shows results obtained in experiments with the following compounds: oxalacetate,  $\alpha$ -ketoglutarate, fumarate, formate, glycerol, lactate, pyruvate and propionate. It is not possible to draw any definite conclusions from these results, but it is interesting to note that oxalacetate inhibited the growth of *Streptococcus thermophilus* strains, and that calcium ions counteracted this inhibition to a certain extent. This particular finding has been confirmed subsequently.

The promotion of growth by calcium and other chemical compounds in the media is being investigated. The experiments carried out to date have

shown that no reciprocal influence exists between calcium and certain sugars such as sucrose, glucose and lactose. It was found that some strains of *Streptococcus thermophilus* cannot utilize glucose; all the strains utilized sucrose and lactose.

Table 15.

Effect of calcium in media containing sodium phosphate. Incubation at 42°.

Basal medium	Phosphate $\mu\text{mole/ml}$	Strain	$\mu\text{g Ca}^{++}/\text{ml}$					Incubation hours
			0	5	10	20	40	
CI	50	Str 10	78	173	141	149	142	42
»	50	Str 10	206	131	128	130	153	42
»	75	Str 10	260	288	241	162	216	42
CIV	50	Str 10	—	—	6	10	33	48
»	75	Str 10	—	—	—	5	5	48
CI	50	Str 74	10	10	86	177	256	48
»	50	Str 74	—	208	273	246	238	48
»	75	Str 74	315	220	220	250	280	48
CIV	50	Str 74	—	—	—	10	31	48
»	50	Str 74	—	—	—	—	11	48
»	75	Str 74	—	—	—	—	—	48
CI	50	Str 75	12	185	32	26	262	48
»	50	Str 75	131	226	206	214	210	48
»	75	Str 75	—	62	220	208	167	48
CIV	50	Str 75	5	—	5	17	33	48
»	50	Str 75	—	—	—	—	11	48
»	75	Str 75	—	—	—	—	—	48
CI	50	Str 101	60	134	136	133	134	48
»	50	Str 101	85	100	105	123	123	48
»	75	Str 101	—	18	216	6	124	48
CIV	50	Str 101	—	—	6	38	35	48
»	50	Str 101	—	—	—	—	9	48
»	75	Str 101	—	—	—	—	—	48
CI	50	Cas J (7)	236	240	109	140	141	48
»	50	Cas J (7)	212	220	202	153	111	48
»	75	Cas J (7)	258	296	298	280	125	48
CIV	50	Cas J (7)	5	281	299	330	335	48
»	50	Cas J (7)	—	—	8	16	20	48
»	75	Cas J (7)	—	—	—	—	—	48



Table 16.

Effect of sodium acetate in medium CI containing glycerophosphate, potassium phosphate, succinate, and tartrate (5  $\mu$ mole/ml of each), tested with and without calcium. Incubation at 42°.

$\mu$ g Ca <sup>++</sup> /ml	Strain	$\mu$ mole/ml						Incubation hours
		0	5	10	20	50	100	
—	Th	31	79	94	97	101	—	24
5	Th	124	144	149	163	167	149	24
—	Str 77	—	78	76	92	87	7	70
5	Str 77	5	75	76	92	87	7	24
—	Str 10	43	50	53	57	70	66	24
5	Str 10	60	68	65	73	78	82	24
—	Hatt.s.	51	92	85	58	68	85	24
5	Hatt.s.	67	105	77	70	54	61	24
—	Lb 80	142	177	222	228	258	294	70
5	Lb 80	172	180	208	246	272	284	70

Table 17.

Effect of malic acid. Incubation 48 hours at 42°.

Basal medium	Strain	$\mu$ mole/ml			
		0	50	100	150
CICa <sup>1</sup> )	Th	20	—	—	—
BI	Th	30	10	—	—
BII	Th	90	—	—	—
CICa	Str 77	10	10	—	—
BI	Str 77	20	10	—	—
BII	Str 77	—	—	—	—
CICa <sup>1</sup> )	Hatt.s.	80	170	170	20
BI	Hatt.s.	70	10	—	—
BII	Hatt.s.	240	200	—	—
CICa	Lb 80	70	120	70	100
BI	Lb 80	70	220	230	200
BII	Lb 80	400	230	180	190

**Table 18.**

Effect of malonic acid. Incubation 48 hours at 42°.

Basal medium	Strain	$\mu\text{mole/ml}$				
		0	50	100	150	200
BI	Th	30	—	—	—	—
BII	Th	10	—	—	—	—
BI	Str 77	20	—	—	—	—
BII	Str 77	40	—	—	—	—
BI	Str 10	50	20	—	—	—
BII	Str 10	70	—	—	—	—
BI	Hatt.s.	170	30	—	—	—
BII	Hatt.s.	240	200	—	—	—
BI	Lb 80	80	220	180	50	10
BII	Lb 80	260	240	190	—	—

**Table 19.**

Effect of high succinic acid concentrations. Incubation 48 hours at 42°.

Basal medium	Strain	$\mu\text{mole/ml}$				
		0	50	100	150	200
CICa <sup>1</sup> )	Th	10	90	—	—	—
BI	Th	40	—	—	—	—
BII	Th	—	—	—	—	—
CICa	Str 77	10	150	—	—	—
BI	Str 77	20	—	—	—	—
BII	Str 77	—	—	—	—	—
CICa	Str 10	10	120	50	—	—
BI	Str 10	20	140	200	—	10
BII	Str 10	60	—	—	—	—
CICa	Hatt.s.	80	170	170	20	—
BI	Hatt.s.	100	210	10	—	—
BII	Hatt.s.	220	180	—	—	—
CICa	Lb 80	70	90	120	110	20
BI	Lb 80	80	200	180	160	30
BII	Lb 80	260	210	150	140	—

Table 20.

Effect of low succinic acid concentrations in medium CI containing acetate, glycerophosphate, and potassium phosphate (5  $\mu$ mole/ml of each), tested with and without  $\text{Ca}^{++}$ . Incubation at 42°.

$\mu\text{g}$ $\text{Ca}^{++}/\text{ml}$	Strain	$\mu\text{mole}/\text{ml}$						Incubation hours
		0	5	10	20	50	100	
—	Th	78	94	95	95	—	—	24
10	Th	115	139	157	169	196	—	24
—	Str 77	26	14	—	—	—	—	24
10	Str 77	67	79	82	87	13	—	24
—	Str 10	51	55	72	82	77	—	24
10	Str 10	54	65	71	98	105	—	24
—	Hatt.s.	89	99	125	76	53	—	24
10	Hatt.s.	95	88	108	65	61	20	24
—	Lb 80	133	191	200	226	272	—	70
10	Lb 80	135	187	224	248	264	264	70

Table 21.

Effect of tartaric acid. Incubation at 42°.

Basal medium	Strain	$\mu\text{mole}/\text{ml}$								Incubation hours
		0	5	10	20	50	100	150	200	
BI	Th	40	—	—	—	—	—	—	—	48
BII	Th	—	—	—	—	—	—	—	—	48
CICa <sup>1)</sup>	Th	20	—	—	—	—	—	—	—	48
CIb	Th	96	105	88	19	—	—	—	—	24
CIbCa	Th	118	132	141	142	79	—	—	—	24
BI	Str 77	10	—	—	—	—	—	—	—	48
BII	Str 77	—	—	—	—	—	—	—	—	48
CICa	Str 77	10	—	—	—	110	—	—	—	48
CIb	Str 77	59	37	—	—	—	—	—	—	24
CIbCa	Str 77	98	101	100	78	—	—	—	—	24
BI	Str 10	30	—	—	—	10	—	—	—	48
BII	Str 10	60	—	—	—	—	—	—	—	48
CICa	Str 10	10	—	—	—	30	—	—	—	48
CIb	Str 10	66	63	65	54	—	—	—	—	24
CIbCa	Str 10	63	77	63	65	—	—	—	—	24
BI	Hatt.s.	100	—	—	—	210	10	—	—	48
BII	Hatt.s.	210	—	—	—	190	—	—	—	48
CICa	Hatt.s.	10	—	—	—	40	—	—	—	48
CIb	Hatt.s.	130	142	142	130	71	—	—	—	24
CIbCa	Hatt.s.	131	126	117	123	93	25	—	—	24
BI	Lb 80	80	—	—	—	200	180	160	30	48
BII	Lb 80	260	—	—	—	200	170	130	10	48
CICa	Lb 80	70	—	—	—	120	120	130	10	48
CIb	Lb 80	158	176	185	202	216	—	—	—	24
CIbCa	Lb 80	159	183	206	218	234	185	—	—	24

Table 22.

Effect of glycerol-2-phosphate in medium CI containing 5  $\mu$ mole/ml each of acetate, potassium phosphate, succinate, and tartrate. Incubation at 42°.

$\mu$ g Ca <sup>++</sup> /ml	Strain	$\mu$ mole/ml						Incubation hours
		0	5	10	20	50	100	
—	Th	34	87	97	—	—	—	24
10	Th	94	139	156	194	18	—	24
—	Str 77	26	11	—	—	—	—	24
10	Str 77	68	86	100	68	—	—	24
—	Str 10	23	50	82	130	71	—	24
10	Str 10	44	69	91	141	210	—	24
—	Hatt.s.	98	101	74	73	34	—	24
10	Hatt.s.	87	98	53	61	28	—	24
—	Lb 80	—	178	214	218	280	—	70
10	Lb 80	145	198	210	230	298	—	70

Table 23.

Effect of malic acid, malonic acid, succinic acid, and tartaric acid in medium CI. Incubation 48 hours at 42°.

Strain	malic acid, malonic acid, succinic acid, tartaric acid, $\mu$ mole/ml of each				
	0	12.5	25	37	50
Th	—	20	50	50	60
Str 77	—	10	40	50	—
Hatt.s.	50	90	90	130	120
Lb 80	60	130	160	160	100

Table 24.

Effect of sodium citrate in the media CI, CII, and CIV. Incubation 48 hours at 37°.

Basal medium	Strain	$\mu$ mole/ml									
		0	2	4	10	15	20	27	34	48	61
CI	Th	61	20	24	—	—	—	—	—	—	—
CII	Th	109	95	48	—	—	—	—	—	—	—
CIV	Th	186	176	—	—	—	—	—	—	—	—
CI	Str 77	—	—	—	—	—	—	—	—	—	—
CII	Str 77	126	95	80	—	—	—	—	—	—	—
CIV	Str 77	138	89	—	—	—	—	—	—	—	—
CI	Str 10	—	17	21	—	—	—	—	—	—	—
CII	Str 10	67	59	68	—	—	—	—	—	—	—
CIV	Str 10	164	142	87	—	—	—	—	—	—	—
CI	Hatt.s.	76	98	126	128	49	120	—	—	—	—
CII	Hatt.s.	193	192	183	6	—	—	—	—	—	—
CIV	Hatt.s.	272	180	127	7	—	—	—	—	—	—
CI	Lb 80	65	88	133	141	164	138	129	26	—	—
CII	Lb 80	176	142	139	122	43	4	—	—	—	—
CIV	Lb 80	280	248	198	192	137	—	—	—	—	—

Table 25.

Effect of oxalacetate (= Oa),  $\alpha$ -ketoglutarate (= Kg), fumarate (= Fu), formate (= Fo), glycerol (= G), lactate (= La), pyruvate (= Py), and propionate (= Pr) in medium CI, tested with 10  $\mu\text{g}$   $\text{Ca}^{++}/\text{ml}$  and without calcium. Incubation 12 hours at 37°.

Compound	Strain	without $\text{Ca}^{++}$ , $\mu\text{mole}/\text{ml}$						with $\text{Ca}^{++}$ , $\mu\text{mole}/\text{ml}$					
		0	5	10	20	50	100	0	5	10	20	50	100
Oa	Lb 80	103	—	—	—			108	3	—	—		
Kg	Lb 80	79	117	130	167			105	130	167	174		
Fo	Lb 80	81	77	73	49	30	33	106	100	79	72	59	46
Fu	Lb 80	73	150	150	124	112	116	117	163	168	135	126	122
La	Lb 80	74	94	105	117	115	70	85	85	126	134	113	88
Py	Lb 80	90	73	58	47	43	20	109	89	78	99	93	38
Pr	Lb 80	83	125	143	146	212	135	98	136	150	146	151	135
G	Lb 80	72	75	74	78	78	82	97	95	94	96	87	89
Oa	Hatt.s.	80	—	—	—			65	—	—	—		
Kg	Hatt.s.	68	99	130	140			73	100	137	127		
Fo	Hatt.s.	74	68	66	65	60	45	77	64	67	68	68	53
Fu	Hatt.s.	58	118	145	173	158	35	80	131	155	173	161	146
La	Hatt.s.	66	78	86	86	21	—	75	82	87	90	27	8
Py	Hatt.s.	71	69	69	58	46	36	69	72	63	62	52	37
Pr	Hatt.s.	67	97	123	153	98	20	71	103	123	134	149	25
G	Hatt.s.	68	65	67	68	68	68	68	69	72	69	69	67
Oa	Str 10	23	49	—	—			31	65	97	—		
Kg	Str 10	32	33	41	68			36	34	55	76		
Fo	Str 10	28	22	17	19	15	9	36	30	19	20	12	9
Fu	Str 10	14	19	21	36	24	—	35	38	44	57	73	64
La	Str 10	29	22	16	13	—	—	34	32	24	20	—	—
Py	Str 10	17	18	24	40	86	40	28	39	53	67	84	45
Pr	Str 10	14	36	47	47	19	4	26	46	49	56	19	8
G	Str 10	13	13	15	16	13	18	24	20	18	21	21	22
Oa	Th	16	—	—	—			33	75	59	—		
Kg	Th	26	40	75	68			37	54	74	91		
Fo	Th	21	31	29	25	14	7	37	47	40	28	18	14
Fu	Th	10	13	15	16	28	—	—	—	—	—	5	—
La	Th	24	19	22	15	—	—	43	49	29	4	—	—
Py	Th	15	15	78	33	97	117	—	—	4	3	6	—
Pr	Th	14	29	31	18	—	—	40	65	72	54	9	—
G	Th	11	10	9	10	12	12	26	26	25	28	28	28

## Discussion

Our investigations have mainly dealt with those lactic acid bacteria whose growth was affected by calcium ions. Very little is known about the effect of these ions on the growth of lactic acid bacteria <sup>9</sup>, and therefore the aim of the experiments was to determine the calcium requirements of the strains studied, and to clarify the extent to which other compounds of the basal media affect these requirements and the uptake of calcium. It was found that calcium ions could not be replaced by any of the other metal ions investigated ( $Mn^{++}$ ,  $Mg^{++}$ ,  $Fe^{++}$ ,  $Ba^{++}$ ,  $Sr^{++}$ ). The experiments also threw light upon the effects of different concentrations of metal ions on the growth of various strains. Only magnesium ions seemed to be essential to most of the strains.

The inhibition of growth by citrate can be counteracted or at least weakened by calcium. It is of interest that the required concentrations of calcium are very low compared to the citrate concentrations. On the basis of this finding and the fact that calcium also counteracts the inhibition of growth by oxalacetate one can conclude that the counteracting effect of calcium ions is not limited to citrate alone. Moreover, it is evident that in all the basal media used calcium ions have a growth promoting effect on the strains.

The fact that the growth of about 30 of the 124 strains of lactic acid bacteria investigated were affected by calcium ions must be considered a finding of great interest. As yet, however, no conclusions can be drawn about the importance of calcium ions in the metabolism of lactic acid bacteria. The results described above indicate that calcium may play a fundamental part in the metabolism of certain organic acids.

Among the 124 investigated strains of lactobacilli there were a few which require unidentified growth factors. As yet nothing can be said about the chemical nature of these growth promoting factors although they have been separated from yeast extract and tomato juice.

The role of calcium in the metabolism of the lactic acid bacteria will be investigated further and the established unidentified growth factors will be studied to determine their chemical nature.

## Summary

The nutritional requirements of 124 strains of lactic acid bacteria isolated as pure cultures from various dairy products have been studied. Some of these strains are used in cheesemaking in Finland. By growing the strains first in non-synthetic and then in synthetic media it was found that many

of them possessed very complex growth factor requirements. Several of the strains required unidentified growth factors. Particularly the *Streptococcus thermophilus* strains require calcium for growth. It is of interest that calcium ions counteract the inhibition of growth by citrate and some other organic anions. In order to make possible a further investigation of the effect of calcium, the efforts were largely concentrated on a search for new nutrient solutions with the simplest possible compositions. Such basal media, the chemical compositions of which must, of course, be known exactly, must first be found before a detailed examination of the mechanism of action of calcium can be undertaken.

#### Acknowledgement

This work is part of a research supported by grant No. FG-Fi-102-60 from the U.S. Department of Agriculture, Washington.

### References

1. KNIGHT, B. C. J. G., *Vitamins and Hormones* 3, (1945) 108.
2. PETERSON, W. H. and PETERSON, M. S., *Bacteriol. Revs.* 9, (1945) 49.
3. SNELL, E. E., *Wallerstein Labs Commun* 11, (1948) 81.
4. —»— *Bacterial Physiology*, ed. Werkman, C. H. and Wilson, P. W., Acad. Press, New York, 1951, p. 214.
5. —»— *Bact. Revs.* 16, (1952) 227.
6. NURMIKKO, V., *Symbiosis Experiments Concerning the Production and Biosynthesis of Certain Amino Acids and Vitamins in Associations of Lactic Acid Bacteria*, (Diss.) *Ann. Acad. Sci. Fennicae, Ser. A. II. No. 54*, 1954.
7. RITTER, W., *VIth International Congress for Microbiology, Rome, Symposium on Nutrition and Growth Factors*, 1953, p. 157.
8. ANDERSON, A. W. and ELLIKER, R. P., *J. Dairy Sci.* 36, (1953) 161.
9. MÄLKKI, Y., *Suomen Kemistilehti B* 32, (1959) 34.



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